Improving Electrostatics Description In Scoring Functions -Insights For Their Role For Drugs

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On importance of electrostatic interactions

- 1. physics: electrostatic interactions are important
- wide practical application within the force fields (FF) and scoring functions (SF) confirms it
- 3. so accurate electrostatics seem to be crucial for SFs...

DENR - a platform for partial charge calculation

- 1. we have developed charges for long time...
- 2. our current and perhaps the best project is empirical method based on the Dynamic ElectroNegativity Relaxation (DENR) principle (incomplete EN equalization) [1]
- 3. our target metrics is quantitative reproduction of the QC reference MEP by DENR charges, so the charges are widely applicable by design
- 4. practical benefits of DENR
 - a. reasonable QC MEP reproduction
 - b. empirical foundations the model is rather "rough", so the chances to overfit are low
 - c. fast calculation (compared to semi-empirical and ab initio QC methods)
 - d. graph based no need for 3D geometry
- 5. our plan (was)
 - a. apply our charges in SF and finally in docking
 - b. we need to first check how it performs





1. Shulga, D. A. et al // SAR and QSAR in Env. Res., 2008, 19(1-2), 153-165.

Experiment design: scoring function (SF)

AutoDock 4.2

- 1. force-field (FF) (or physics) based SF
- 2. widely used and freely accessible
- 3. already uses Gasteiger partial charges to describe electrostatic part of interactions
- 4. there are two papers in which better quality MEP reproductions charges were tested in combination with AD4.2
 - a. RESP (gold standard), AM1-BCC (practical standard) [1]
 - b. several known charge methods [2]
- 5. in common
 - a. promising results reported for QC MEP related methods (RESP and AM1-BCC) [1]
 - b. ...but low throughput and high complexity are the price
 - 1. Wang J. C. et al. // JCIM, **2011**, 51, 2528-2537.
 - 2. Hou X. et al. // JCIM, **2013**, 53, 188-200.

Experiment design: ligand-receptor complex set

Δ

 ligand-receptor complex set - core set from CASF 2016 Update [1] - 285 structures

 $+ W_{tor} \times N_{tors}$

- 2. obtain the best linear model for each charge set (ligand and receptor)
- 3. AutoDock4.2 score:
- task: obtain new set of Wj for each charge scheme

$$egin{aligned} G_{ ext{bind}} &= W_{ ext{vdW}} imes \sum_{i,j} \left(rac{A_{ij}}{r_{ij}^{12}} - rac{B_{ij}}{r_{ij}^6}
ight) + W_{ ext{H-bond}} \ & imes \sum_{i,j} E(t) \left(rac{C_{ij}}{r_{ij}^{12}} - rac{D_{ij}}{r_{ij}^{10}}
ight) \,+ \, W_{ ext{estat}} imes \sum_{i,j} rac{q_{id}}{arepsilon(r_{ij})} \ &+ \, W_{ ext{desol}} imes \sum_{i,j} \left(S_i V_j \,+ \, S_j V_i
ight) \, ext{e}^{(-r_{ij}^2/2\sigma^2)} \end{aligned}$$

NB: Desolvation values taken using Gasteiger charges for all other charges

1. Su M. et al. // JCIM, **2018**, 59, 895-913.



Experiment aim in short

Study whether replacement of Gasteiger charges (default in AutoDock4.2) with our DENR charges (with parameters fit to QC MEP) will brings benefits (which we expect)

All in fair conditions!

Results

The main outcome

Gasteiger (L) - Gasteiger (R)

DENR (L) - DENR (R)



Improvement is negligible!

The main outcome



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The questions arising

- 1. technical
 - a. a purely empirical HB term inside physics (FF) based SF possibly duplicating electrostatics
 - b. desolvation is very crude there's room for improvement
- 2. conceptual
 - a. is electrostatics important at all? when we consider small molecule receptor interactions in solution?

Electrostatics in scoring ligand-receptor interactions

dilemma

- a. on one hand,
 - i. in (vacuum) physics E(electrostatic) is a dominating and long range interaction
 - ii. it is omnipresent in practice there must be firm reason for it!
- b. on the other hand,
 - i. E(electrostatic) contribution is negligible in FF-based score (in energy terms)
 - ii. wide variation of charges different E(electrostatic) quality does not affect the outcome much

Electrostatics in scoring ligand-receptor interactions

further thoughts

- a. main driving forces of interactions
 - i. hydrophobic interactions
 - ii. shape complementarity
- b. decoration: HB as directed interactions
- c. BUT!
 - i. thermodynamic effect of HB is highly questionable on affinity!
 - ii. L · N(H2O) + R · M(H2O) → L-R (K HBs)
 + (M+N) · H2O, K << M+N (geometrically)





Taken from Med. Chem. Commun., 2017, 8, 1970

Electrostatics in scoring ligand-receptor interactions

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 - ii. L · *N*(H2O) + + (*M*+*N*) · H2



Fig. 1 Frequency distribution of the most common non-covalent interactions observed in protein-ligands extracted from the PDB.

Taken from Med. Chem. Commun., 2017, 8, 1970

What we actually seek for?



Highly specialized tool *Ideal* for *just* L-R affinity prediction!

What we actually seek for?



Highly specialized tool *Ideal* for *just* L-R affinity prediction!



VS

Versatile tool **Good enough** for prediction of *overall* success ligand -> drug!

Idea elaboration: HBs beyond affinity in SFs

Hypothesis

a. HBs and electrostatics serve not for the affinity reasons! But rather for overall "success"

Reasons

- a. datasets (PDBBind, etc.) for SF development contain (almost) drugs as positive examples -> other properties crucial for drugs were taken indirectly into account = "Survivorship bias"
- b. presence of HBs (HBD, HBA) in L-R complex
 - i. directed interactions, demand for complementarity off-target selectivity
 - ii. if in desired range (e.g. Ro5) favorable ADMET properties
 - iii. improved synthesizability

Idea elaboration: HBs beyond affinity in SFs

Graphic illustration

(Maximum likelihood principle)

To get a drug one must obtain both high affinity and good drug-like properties...



Claims and consequences

Reasons

a. Claim: most contemporary scoring functions (SFs) measure not only pure **affinity**, but rather a more complex function estimating "**ligand developability**"

- b. Corollary: these different contributions should be explicitly split and consciously dealt with
 - i. sub-corollary: directed electrostatic interactions and hydrogen bonds (HBs) are more relevant to estimate the relevant ADMET properties, rather than the affinity per se (where it seems to be useless!)

Thank you!